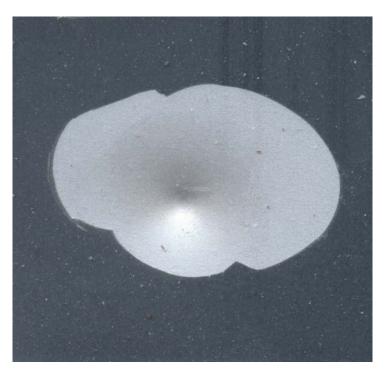
# TU Clausthal

### ENAMELLING WITH REDUCED FRACTIONS OF COBALT AND NICKEL: EFFECT OF SURFACE ROUGHNESS ON ENAMEL ADHESION

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#### **MOTIVATION**

#### Aim of the project:

Approach for eliminating the use of adherence oxides of cobalt and nickel in ground and direct enamel

#### **Experimental approach:**

- I. Pretreatment of the steel surface (mechanical and chemical methods)
- II. Investigations of the processes during firing conventional enamels
- III. Alternative adherence oxides and mixtures thereof (electrochemically similar to Ni and Co)
- IV. Variation of the enamel composition (thermal application window)



#### PRETREATMENT METHODS AND ENAMELLING

- Steel sheet (d = 1.6 mm, I = 100 mm, low-carbon, Type YMVIT ULTRA, DC04ED, Tata Steel (NL))
- Degreasing
- Pretreatment procedure
  - A: Sandblasting (Corundum, steel shot, chilled cast iron)
  - B: Etching in aqueous FeCl<sub>3</sub> solution (c = 2 mol/l: T = 20°C / T = 70°C for 1 h)
  - C: TEOS (Immersion in TEOS solution, sol-gel-chemistry route "Tetra-Ethyl-Ortho-Silane", drying at 20 °C)
- Enamel: standard ground enamel (Wendel-Email, Germany) and mixtures thereof until Co-/Ni-free (Co100 Co75 Co50 Co25 Co0, containing adherence oxides from 100% to 0%)
- Applying the enamel: Spraying (layer thickness 200-300 μm), drying at 80 °C / 30 min.
- Firing (Parameters: T = 800 °C, t = 7 minutes, electrical furnace)
- Cooling (air)
- Testing the adherence by several methods → deep drawing (cupping test)
- Quantitative examination of the damage (scanning the surface, picture analysis via ImageJ (imagej.nih.gov/ij))



# METAL PRETREATMENT: BLASTING MATERIALS

#### Steel shot



#### Chilled cast iron



#### Corundum

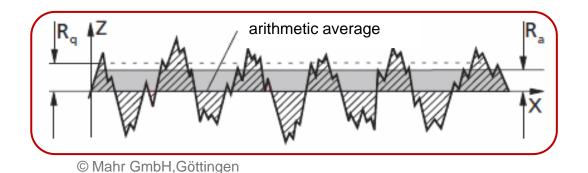




#### DETEKTION OF ROUGHNESS PARAMETERS



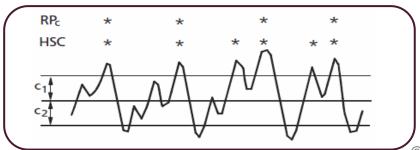
Mobile roughness measuring unit M300, Fa. Mahr GmbH, Göttingen



Arithmetic average,  $R_a$ : the mean distance of a point on the surface from the profile center line

#### **Profile roughness parameters:**

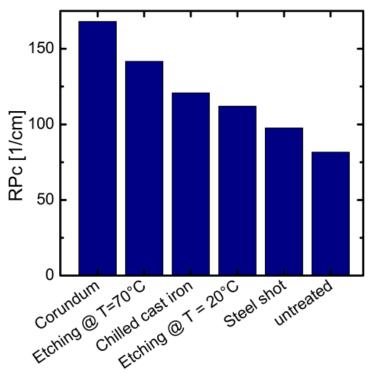
$$\mathbf{R_a}$$
,  $\mathbf{R_g}$ ,  $\mathbf{R_z}$ ,  $\mathbf{R_p}$ ,  $\mathbf{R_v}$ ,  $\mathbf{R_k}$ ,  $\mathbf{RP_c}$ ,  $\mathbf{R_{sm}}$ ...



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Peak count,  $\mathbf{RP_c}$ : the number of profile elements per cm that exceed a certain average level  $c_1$  and then fall below the lower  $c_2$ 

#### PRETREATMENT RESULTS: ROUGHNESS PARAMETERS



 $R_{a}$  [ $\mu m$ ] Etching @ T = 20°C Steel shot

Profile peak number RPc as a function of pretreatment procedure

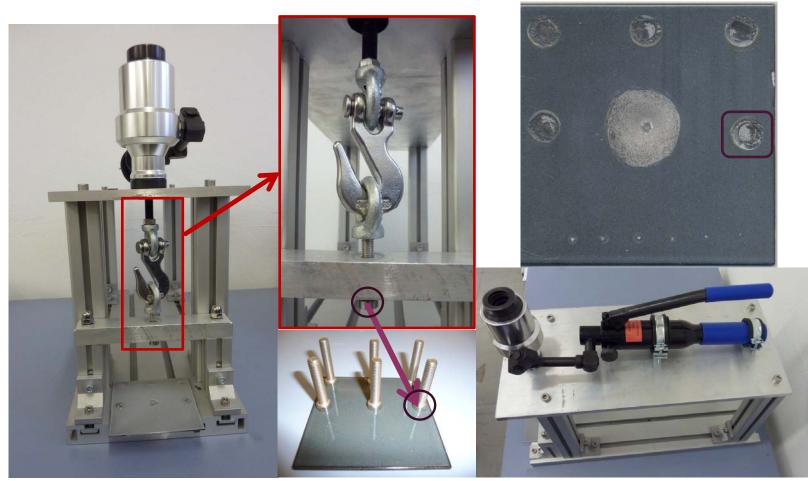
Surface roughness Ra as a function of pretreatment procedure

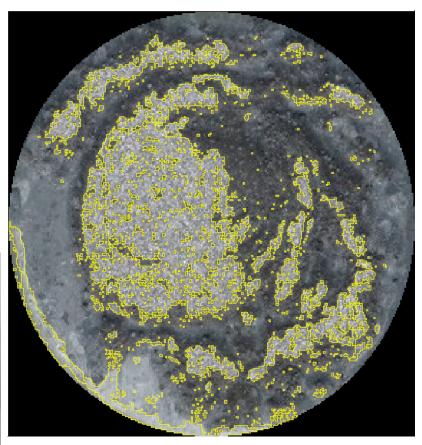
- Increase of peak counts from 80 (untreated) up to 170 cm<sup>-1</sup> (Corundum blasted)
- Highest mean roughness at about 4.5 µm (chilled cast iron)
- Etching leads to significant higher roughness parameters
  - > RP<sub>c</sub> values 142 resp. 112 cm<sup>-1</sup>
  - > R<sub>a</sub> values of 2.9 resp. 2.2 μm





# ADHERENCE TESTING PROCEDURE: PULL-OFF TEST





→ disadvantage on steel sheet: uncertain crack initiation and crack propagation, minor differentiation of adherence

## ADHERENCE TESTING PROCEDURE: DEEP DRAWING (CUPPING TEST)



- Deep drawing in a sheet metal testing machine
- Access to adherence by means of open metal surface fraction x (<10%: good adhesion)</li>
- Adherence index a = 100 x
- → Advantage: better differentiation of adherence status



### RESULTS OF PRETREAMENT

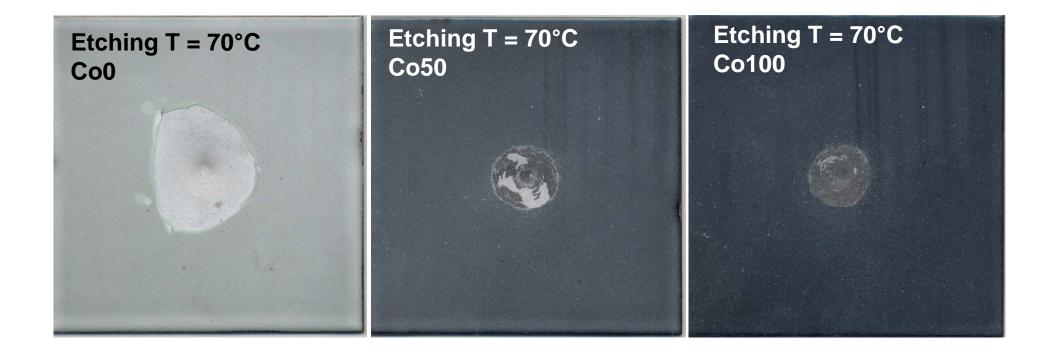
- **Coating on steel sheet:** 
  - Coating with TEOS leads to a dramatic bad adherence (no adherence promoting layer)
- Possible causes:
  - Surface roughness is decreased due to the coating layer of SiO<sub>2</sub>
  - Filling of the cavities decreases the number of anchor-points
- SiO<sub>2</sub> layer: barrier for Oxygen inhibits oxidation of the steel surface and Fe<sup>2+</sup> migration into the melt





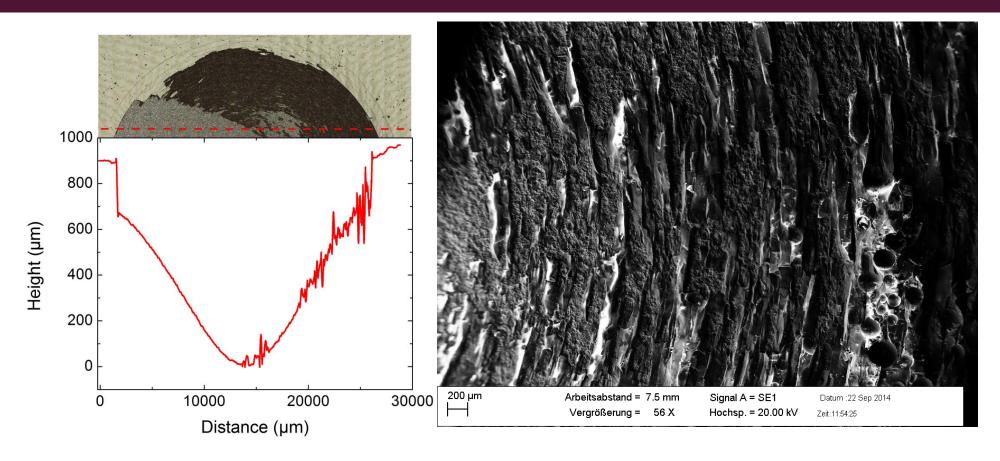


## RESULTS OF PRETREAMENT: ETCHING



- Rough surface supports adherence due to an increase of mechanical interlocking
- Without CoO/NiO: no material compound can be achieved, no chemical bonding reaction

## DAMAGE ASPECT: INDICATOR FOR ADHERENCE BEHAVIOR



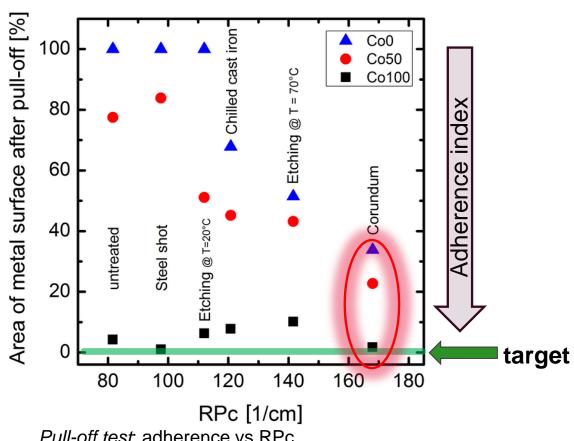
- Steep and plain edge on left side: less adherence, absence of dark colored interface
- "Flaked" scaling (right side) of enamel layer points out a higher/better adherence behavior, presence of a dark colored interface, high iron content



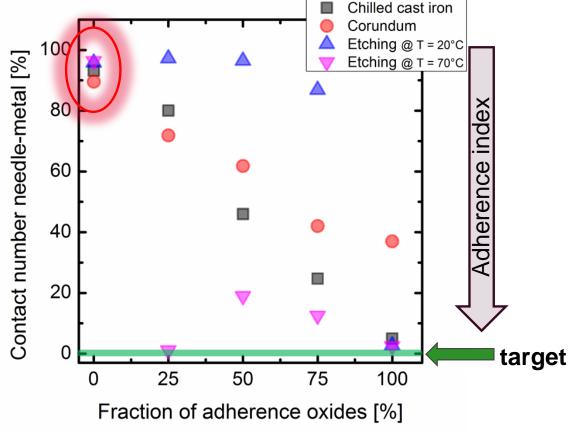




#### RESULTS OF PRETREATMENT ON ADHERENCE INDEX



Pull-off test: adherence vs RPc, parameter: adherence oxide content



Deep drawing test: adherence vs fraction of adherence oxide, parameter: Pretreatment

25/5/2015

→ Advantage of deep drawing test procedure versus pull-off test





#### DEEP DRAWING TEST PROCEDURE VERSUS PULL-OFF TEST

#### Results from method Pull-off test

- Higher surface roughness can provide a better adherence with an increase up to 60% using vitreous enamel being completely free of cobalt and nickel (blasting medium Corundum)
- Effect of the chemical component (presence of adherence oxide) is stronger, no difference in adherence if comparing plain to rough steel surfaces

#### Results from method Deep drawing

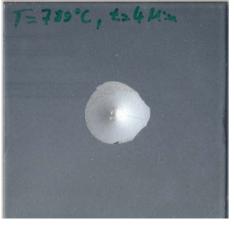
- Only very small differences in adherence index between roughened and untreated steel surface
- Mechanical roughening increases mechanical interlocking but produces no material compound (i.e. no adherence)
- Most important is the effect of chemical interaction of adherence oxides in the melt



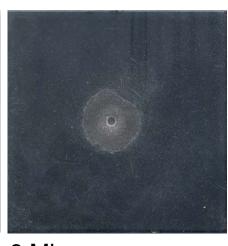
# **ENAMELLING: INFLUENCE OF TEMPERATURE**

**780°C**Standard ground









2 Min

4 Min

6 Min

8 Min

**820°C**Standard ground









2 Min

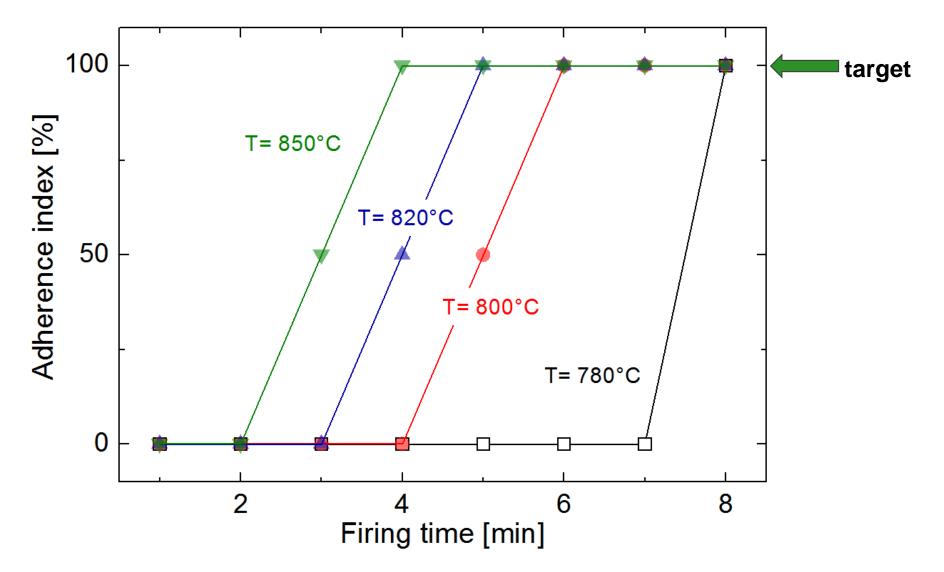
4 Min

5 Min

6 Min



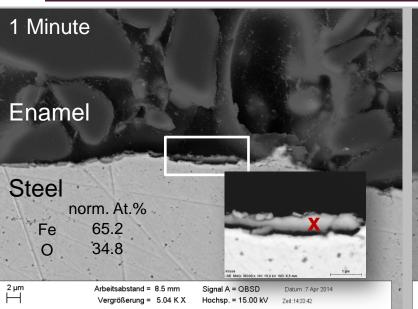
## ADHERENCE ACCORDING TO TIME AND TEMPERATURE

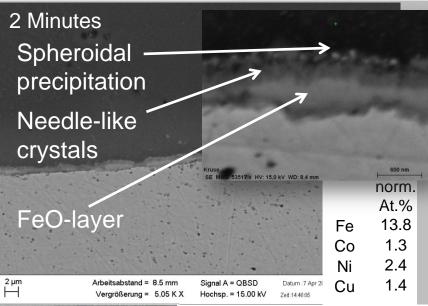


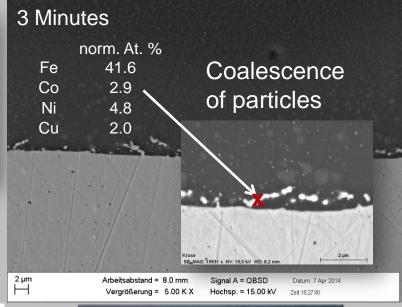




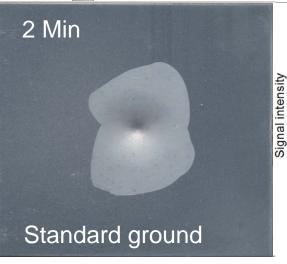
## MECHANISM DURING FIRING OF CONVENTIONAL GROUND ENAMEL

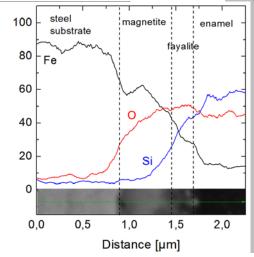


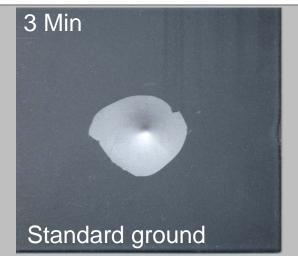




1 Min Standard ground



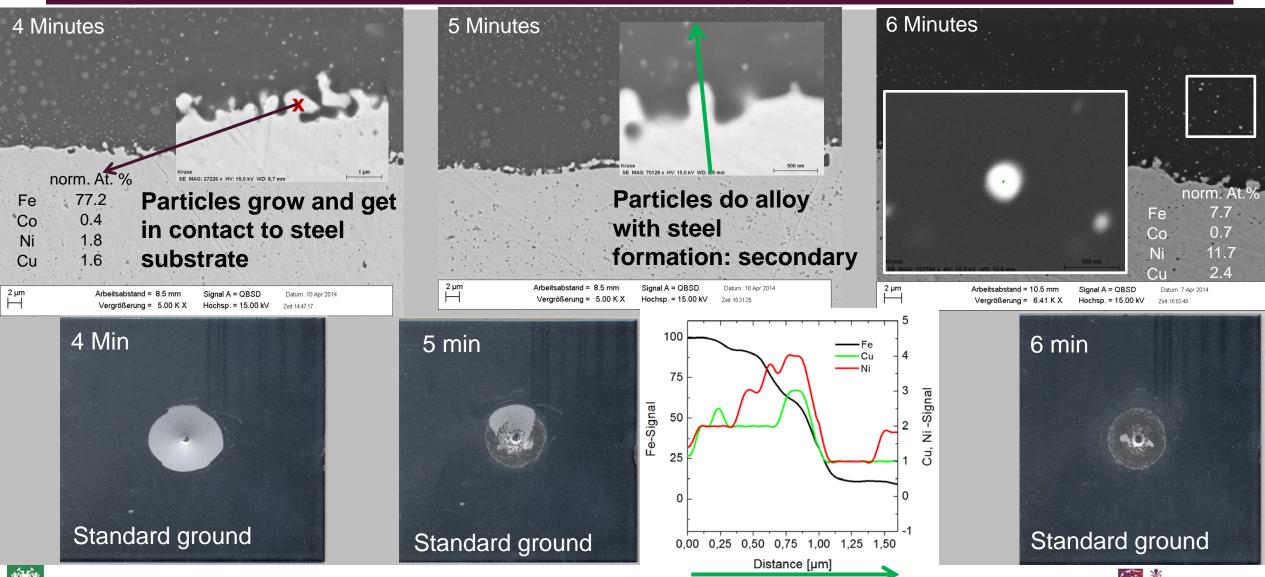








# MECHANISM DURING FIRING CONT.







#### COMPREHENSION OF MECHANISMS DURING FIRING

#### of conventional (adherence oxides bearing) ground enamel

- 1. Oxidation of base metal (steel) in the initial stages of firing (Minute 1 and 2)
- 2. Formation of Fayalite (Fe<sub>2</sub>SiO<sub>4</sub>) from iron oxide and enamel (from Minute 2)
- 3. Dissolution of Fayalite in the glass melt
- 4. (Over-) saturation of iron cations in the glass melt
- 5. Precipitation of iron, forming alloys of Fe-Co-Ni-Cu (metal particles) (from Minute 3)
- 6. Growth/coarsening of micro alloys and subsequently getting in contact with base metal
- 7. Forming of a rough surface on steel sheet (as an **additive** event, no corrosion) → anchor-points
  - → good adherence
- 8. Formation of a material compound with high surface roughness (from Minute 5)





### RESULTS OF EXPERIMENTAL RESEARCH

I. Pretreatment of the steel surface (mechanically and chemically)

Good interlocking of the enamel with the steel sheet surface but

→ no formation of a material compound without adherence oxides

II. Significance of testing procedures for the adherence of enamel on steel sheet

Results and statements depend on the testing procedure: deformation processes provide a better differentiation of the adherence index (enamel on metal) than pull-off or impact tests

III. Studies of the processes during firing of conventional enamels

Contribution to the clarification of a core aspect of the formation of metal-enamel compounds

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